



Uncertainty

What it is, where we need it, how does uncertainty propagate

What is uncertainty?

- ▶ A police says your car was travelling at a speed of 70 km/h. What does he really mean? Now he has to give you a better answer. He says you were speeding because the speed limit is 50 km/h and
the speed of your car was (70 ± 5) km/h
 - ▶ 70 km/h is the **value** of the speed, also called the best estimate: v ;
 - ▶ 5 km/h is the **uncertainty** of the speed: δv ;
- ▶ The **value** and the **uncertainty** together define a **range** (65 to 75 km/h) where the “true” speed most likely falls within. i.e.: If someone were taking the same measurement, he/she would likely get a value within this range.
- ▶ How likely? It depends. For example, 68% likelihood if using “standard deviation” as the uncertainty for a normal distribution. We will try “70% or more” for our labs.
- ▶ None of above terms is exact: they are all uncertain.

Absolute uncertainty and relative uncertainty

- ▶ Uncertainty that describe the range of the data is also called “the absolute uncertainty”. It has the same units as the value. Symbol is δA .
- ▶ Is it a big uncertainty or a small uncertainty? 1 meter is a big uncertainty for the size of our lab, but is a small uncertainty for the distance to the moon. The precision of the measurement is reflected by the relative uncertainty $\delta A/A$.

Speed of the car v is (70 ± 5) km/h

$$\text{Relative uncertainty } \frac{\delta v}{v} = \frac{5 \text{ km/h}}{70 \text{ km/h}} = 0.0714 \approx 7\%$$

- ▶ Absolute uncertainty has units, but relative uncertainty has no units.
- ▶ Absolute and relative uncertainty: knowing one = knowing them both.



Where is uncertainty in our lab report?

1. Purpose: no
2. Apparatus: no
3. Data: Yes. All data must have uncertainties (except counting a small integer). Decide uncertainty based on (1) measuring devices (2) the situation.
4. Calculations: no
5. Uncertainty Analysis: Yes. Here we show how the uncertainties of the raw data propagate into the uncertainty of the result.
6. Conclusion: Yes. You must report result with uncertainty.
7. Discussions: Yes. Compare the result to a reference number to see if they agree within uncertainty and discuss the sources of uncertainty.



Uncertainty Propagation

No more sig. fig. rules!

Unc. of data \rightarrow Unc. of result

You need

- ▶ The equation that calculates the result
- ▶ Propagation rules

Propagation Rule # 1: addition and subtraction

- If $z = x + y$ or $z = x - y$, then $\delta z = \delta x + \delta y$

Example:



Propagation Rule # 1: addition

► If $z = x + y$, then $\delta z = \delta x + \delta y$

$$n_1 = 130 \pm 3, n_2 = 50 \pm 2, n_3 = 20 \pm 1$$

Total number of the beads:

$$n = n_1 + n_2 + n_3 = 130 + 50 + 20 = 200$$

$$\delta n = \delta n_1 + \delta n_2 + \delta n_3 = 3 + 2 + 1 = 6$$

So the total number of beads is 200 ± 6 .



Propagation Rule # 1: subtraction

► If $z = x - y$, then $\delta z = \delta x + \delta y$

$$n = 200 \pm 2, n_{\text{out}} = 70 \pm 7$$

Number of beads left in the bottle:

$$n_{\text{in}} = n - n_{\text{out}} = 200 - 70 = 130$$

$$\delta n_{\text{in}} = \delta n + \delta n_{\text{out}} = 2 + 7 = 9$$

So the number of beads left in the bottle is 130 ± 9 .



Propagation Rule # 2: Multiplication and division

► If $z = xy$ or $z = x/y$, then $\frac{\delta z}{z} = \frac{\delta x}{x} + \frac{\delta y}{y}$

Example: Adam rides his bike to Langara. He covers (3.2 ± 0.2) km in (10 ± 1) minutes. What is his average speed?

$$v = \frac{d}{t} = \frac{3.2 \text{ km}}{10 \text{ min}} = 0.32 \text{ km/min}$$

$$\frac{\delta v}{v} = \frac{\delta d}{d} + \frac{\delta t}{t} = \frac{0.2}{3.2} + \frac{1}{10} = 0.1625$$

$$\delta v = \frac{\delta v}{v} \times v = 0.1625 \times 0.32 \text{ km/min} = 0.052 \text{ km/min} \approx 0.05 \text{ km/min}$$

So his average speed is (0.32 ± 0.05) km/min ($\pm 16\%$).

Propagation Rule # 3: Multiplication by a constant

► If $z = kx$ then $\delta z = k\delta x$, or $\frac{\delta z}{z} = \frac{\delta x}{x}$

Example: Betty bought 10 bottles of beads online. Each bottle contains 200 ± 2 beads, which is a 1% uncertainty. For 10 bottles, it will be 2000 ± 20 beads, which is also a 1% uncertainty. Multiplying by 10 increases the absolute uncertainty from 2 to 20, but does not change the relative uncertainty.

Use absolute uncertainty:

$$\begin{aligned}n &= 10n_1 = 10 \times 200 = 2000 \\ \delta n &= 10\delta n_1 = 10 \times 2 = 20\end{aligned}$$

Use relative uncertainty:

$$n = 10n_1 = 10 \times 200 = 2000$$

$$\frac{\delta n}{n} = \frac{\delta n_1}{n_1} = \frac{2}{200} = 0.01$$

$$\delta n = \left(\frac{\delta n}{n}\right)n = 0.01 \times 2000 = 20$$

Both methods get the same result: 10 bottle of beads will have 2000 ± 20 beads.

Propagation Rule # 4: Powers

► If $z = x^n$ then $\frac{\delta z}{z} = |n| \frac{\delta x}{x}$ Don't mix up exponent and constant!

Example: Adam is painting a square wall. Its length and width are both (2.00 ± 0.05) m.

$$A = a^2 = (2.00 \text{ m})^2 = 4.00 \text{ m}^2$$

$$\frac{\delta A}{A} = 2 \frac{\delta a}{a} = 2 \times \frac{0.05}{2.00} = 0.05$$

$$\delta A = \frac{\delta A}{A} \times A = 0.05 \times 4.00 \text{ m}^2 = 0.2 \text{ m}^2$$

The area of the wall is $(4.0 \pm 0.2) \text{ m}^2$ ($\pm 5\%$).

Apply the rules to many variables

$$z = a + b - c + d \rightarrow \delta z = \delta a + \delta b + \delta c + \delta d$$

$$z = \frac{xy}{uvw} \rightarrow \frac{\delta z}{z} = \frac{\delta x}{x} + \frac{\delta y}{y} + \frac{\delta u}{u} + \frac{\delta v}{v} + \frac{\delta w}{w}$$

$$z = \frac{x^3 y}{4\sqrt{vw}} \rightarrow \frac{\delta z}{z} = 3 \frac{\delta x}{x} + \frac{\delta y}{y} + \frac{\delta 4}{4} + \frac{1}{2} \frac{\delta v}{v} + \frac{\delta w}{w}$$